

Seat No.

HAJ-003-1015005

B. Sc. (Sem.-V) (CBCS) (WEF-2016)

Examination

May - 2023

C-501 : Inorganic and Industrial Chemistry (2018)

(Old Course)

Faculty Code : 003 Subject Code : 1015005

Time : $2\frac{1}{2}$ Hours / Total Marks : 70

Instructions :

- (1) All questions are compulsory.
- (2) In all questions : b, c, d have internal options.
- (3) Each section (a, b, c, d) of a question should be written separately in the answer book.
- (4) Figures to the right indicate full marks of sub questions.

1 (a) Answer the following questions :

- (1) Write the normalized wave equation for particle moving in one dimensional box.
- (2) " θ equation is useful to know about magnetic quantum number." True or False.
- (3) What is linear operator ?
- (4) Give equation of Normalization condition.

(b) Answer in brief : (any one out of two)

- (1) What is commutator of operator ?
- (2) Define zero point energy for one and three dimensional systems.

(c) Answer in detail : (any one out of two)

- (1) Explain Hamiltonian operator.
- (2) Derive relation between cartesian and polar coordinates.

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	(d)	Write a note on : (any one out of two) 5
		(1) Derive an energy equation for a particle moving in one dimensional box
		(2) Calculate the energy of 15 orbital.
2	(a)	Answer the following questions : 4
		(1) Which orbitals are not much affected by ligands?
		(3) Which three d-orbitals are known as $t_{2,2}$ orbitals ?
		(4) $[NiCl_4]^{2-}$ is paramagnetic - True or False.
	(b)	Answer in brief : (any one out of two) 2
		(1) Calculate magnetic moment of $[Ni(Br)_4]^{-2}$ complex ion. (2) Explain high spin and low spin complexes.
	(c)	Answer in detail : (any one out of two) 3
		(1) Explain the orbital contribution to magnetic momentum in various crystal fields
		(2) Explain the splitting of d-orbitals in tetrahedral field.
	(d)	Write notes on : (any one out of two) 5
		(1) Discuss splitting of d-orbitals in octahedral field with CFSE.
		(2) Explain factors affecting splitting energy.
3	(a)	Answer the following questions : 4
		 Define : Metal carbonyls. Dress the structure of Fe (CQ)
		(2) Draw the structure of $Fe_3(CO)_{12}$ (3) Give chemical formula of C AF
		(4) Give the full form of RCC.
	(b)	Answer in brief : (any one out of two) 2
		(1) Explain classification of metal carbonyls. (2) Explain setting and hardening of comput
		(2) Explain setting and hardening of cement.
	(c)	Answer in detail : (any one out of two) 3
		(1) Explain metal introsyls.(2) Give merits and demerits of wet and dry process of manufacturing cement.
	(d)	Write notes on : (any one out of two) 5
		 Explain nature of M-CO bond with spectral support. Discuss about types of cement.
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4	(a)	Answer the following questions :	4
		(1) Give the formula of Chile Saltpeter.	
		(2) Write formula of Biuret.	
		 (3) Give any two names of micro nutrient elements. (4) Define fortilizers 	
		(4) Define fertilizers.	
	(b)	Answer in brief : (any one out of two)	2
		(1) Explain action of urea as fertilizer.	
		(2) Write about essential requirements of fertilizer.	
	(c)	Answer in detail : (any one out of two)	3
	(•)	(1) Explain NPK fertilizer with nomenclature.	•
		(2) Explain action of CaCN₂ as fertilizer.	
	(d)	Write notes on : (any one out of two)	5
		(1) Discuss the Prilling method to manufacture ammonium nitrate	
		(2) Describe manufacturing of mono ammonium phosphate	
		and diammonium phosphate.	
5	(a)	Answer the following questions :	Λ
3	(a)	Answer the following questions : (1) Which compounds are used to decelourize glass ?	4
		(1) Which compounds are used as ovidizing agents in glass?	
		(2) Which substances are used as oxidizing agents in glass:(3) Give formula of Feldspar	
		(4) Define glass.	
	(b)	Answer in brief : (any one out of two)	2
		(1) Write the properties of glass.	
		(2) Explain lead glass.	
	(c)	Answer in detail : (any one out of two)	3
	(0)	(1) Write about colouring agents for glass	0
		(2) Explain glass wool.	
	(d)	Write notes on : (any one out of two)	5
		(1) Discuss the raw materials used in glass manufacturing.	
		(2) Write short notes on :	
		(1) Formation of batch material and	
		(2) Annealing for manufacture of glass.	